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Worksheet: Acids, Bases, and Salts

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Review. advertisement Spring 2-14  
Chemistry Review Material Acids, Bases,  
and Salts Name \_\_\_\_ An Arrhenius acid  
is defined as any compound that  
dissociates in aqueous solution to form  
\_\_\_\_ ions.  $\text{HNO}_3 \rightarrow \text{H}^+ + \text{NO}_3^-$   $\text{HCl (aq)}$   
\_\_\_\_ An Arrhenius base is defined as  
any compound that dissociates in  
aqueous solution to ...

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Class 7 Science Chapter 5: Acid Base  
and Salt. Hope you like them and do not  
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at the end of the page. > Question 1 Fill  
in the blanks (a) The most commonly  
used natural indicator is \_\_\_\_ and is  
extracted from \_\_\_\_ ...

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a) Limewater is the common name for a diluted solution of calcium hydroxide  
c) Range 1-3 d) Strong Base is NaOH and Weak Base is  $\text{NH}_4\text{OH}$  e) A hydrated salt is copper sulphate crystals ( $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ ). f) A universal indicator is a mixed indicator of organic chemicals which not only shows whether the given solution is acidic or basic, but also shows the approximate pH values by giving a wide ...

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### **NCERT Solutions for class 10 chapter 2 Acids, bases and salt**

1. acids 2. citric acid 3. Base 4. calcium sulphate hemihydrate,  $\text{CaSO}_4 \cdot \frac{1}{2}\text{H}_2\text{O}$   
5. yellow colour, pink colour 6. Hydrochloric Acid (HCl) 7. Acetic Acid ( $\text{CH}_3\text{COOH}$ ) 8. Sodium Hydroxide (NaOH) 9. Ammonium Hydroxide ( $\text{NH}_4\text{OH}$ ) 10. sodium carbonate,  $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$ , permanent hardness 11. PH 12. Acid 13. Base 14. Neutral 15. decreases 16 ...

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## **CBSE Class 10 - Science - CH2 - Acids, Bases and Salts ...**

Worksheet: Acids, Bases, and Salts  
Name\_\_\_\_\_ What are Acids, Bases, and Salts? The Swedish chemist Svante Arrhenius introduced the theory of ionization and used this theory to explain much about the behavior of acids and bases. ♦ An Arrhenius acid is defined as any compound that dissociates in aqueous solution to form H<sup>+</sup> ions.

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In a neutralization reaction, acid and base reacts to form salt and water. Heat is evolved in the reaction. Question 4: Choose the odd one. Ammonium hydroxide. Tartaric acid. Ascorbic acid. Formic acid. Correct Option is : 1. Solution : Except ammonium hydroxide, rest all are bases. Question 5: China rose indicator turns basic solutions to ...

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and solutions answer key, Chapter 15  
work 4 acid base properties of.

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Acids, Bases and Salts Class 7 Science  
Extra Questions Short Answer Type  
Questions. Question 1. Write the  
characteristics of acids by which we can  
identify acids. Answer: Characteristics of  
acids are. They are sour in taste. They  
turn blue litmus paper/solution to red.

## **Acids, Bases and Salts Class 7 Extra Questions Science ...**

Sodium hydrogen sulphite,  $\text{NaHSO}_3$   
(Acid salt) Sodium silver cyanide,

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Na[Ag(CN)<sub>2</sub>] (Normal salt/Complex salt)  
Basic lead nitrate, Pb(OH)NO<sub>3</sub> (Basic salt)  
Potassium zincate, K<sub>2</sub>ZnO<sub>2</sub> (Normal salt)  
Alum or potash alum, K<sub>2</sub>SO<sub>4</sub>.Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>.24H<sub>2</sub>O (Normal salt/Double salt)  
Calcium bicarbonate, Ca(HCO<sub>3</sub>)<sub>2</sub> (Acid salt)  
Basic copper chloride, Cu(OH)Cl (Basic salt)

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oxide  $\text{Na}_2\text{O}$   $\text{HMnO}_4$   $\text{Ba}(\text{OH})_2$   $\text{KCl}$   $\text{HF}$

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