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2.4section Chemical Reactions Interactive Reader 1 Teacher Notes and Answers SECTION 4 Instant Replay 1. The bonds of reactants break and then bonds form to produce the products. An exothermic reaction gives off more energy 2. than it uses, and an endothermic reaction uses more energy than it gives off. Vocabulary Check 1. chemical reactions 2 ...

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Interactive Reader and Study Guide 1 The Nature of Life Science SECTION1 Asking About Life The Nature of Life Science Name Class Date CHAPTER 1 After you read this section, you should be able to answer ... answer these questions, some scientists design experi-ments to see what makes plants grow larger or faster.

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The light-independent* reactions use the energy from the light-dependent reactions to make sugars. These reactions happen in the stroma. During this process, carbon dioxide (CO₂) and energy from the light-dependent reactions are used to build sugars, usually glucose (C₆H₁₂O₆). The equation for the whole process of photosynthesis is shown ...

seCTion 4.2 Overview of Photosynthesis

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MCQ Questions for Class 10 Science Chemical Reactions and ...

1. A catalyst for reactions in living things is a(n) . 2. A(n) is a reactant that binds to a catalyst. 3. An enzyme is a kind of . 4. What are two ways a catalyst affects a chemical reaction? 5. A catalyst is not a product or a reactant of a chemical reaction. Why not? 2.5 Vocabulary Check Go back and high-light each sentence that has a vocabulary

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Interactive Textbook 259 Chemical Reactions SECTION 4 Name Class Date Energy and Rates of Chemical Reactions continued How Fast Do Reactions Occur? A chemical reaction happens only if reactant particles collide. Even when they do collide, a reaction can occur only if there is enough energy to break chemical bonds.

CHAPTER 14 SECTION 4 Energy and Rates of Chemical Reactions

Interactive Reader 144 Chemical Reactions SECTION 2 Name Class Date Chemical Equations continued What Are Mole Ratios? Consider again the reaction between magnesium and oxygen to produce magnesium oxide. $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ The balanced equation tells you that two Mg atoms react with one O_2 molecule to produce two MgO molecules.

CHAPTER SECTION 12 Chemical Equations

chemical reactions that make up the Calvin cycle. The chemical reactions of the Calvin cycle use carbon dioxide (CO_2) and energy from ATP and NADPH to make sugars. * Academic Vocabulary channel a passage, or a space that can be moved through Light-independent reactions take place in the stroma. stroma (fluid) Interactive Reader 3

Teacher Notes and Answers

Interactive Reader and Study Guide 52 Properties of Matter SECTION3 Chemical Properties ... Compare In a chemical reaction, how do the chemical properties of the products compare with the chemical properties of the reactants? After you read this section, you should be able to answer

CHAPTER SECTION 3 Chemical Properties

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Answers Chemical Reactions Chapter Tests Chemical Reactions magnesium and oxygen in MgO. Mg Chapter Test A 1. C 2. C 3. A 4. B 5. A 6. C 7. A 8. B 9. C 10. B 11. solid 12. products 13. activation energy 14. bias 15. replacement 16. F, Chemical 17. F, remains the same 18. T 19. F, Objective 20. F, dependent on 21. T 22. A physical change. The ...

Answers Chemical Reactions magnesium and oxygen in MgO. Mg

Chapter 7: Chemical Reactions DISCUSSION PLAN Chapter Summary Chapter 7 reviews the information about chemical changes and explains how a chemical reaction can be written down as a chemical equation. The concept of strong electrolytes is introduced to explain solubility and precipitation reactions. Precipitation reaction is used to demonstrate types of chemical equations: molecular, complete ...

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